

CONCENTRATED NITRIC ACID

SECTION 1 - CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

Chemical Name : Concentrated Nitric Acid

Chemical Formula : HNO₃

CAS Number : 7697-37-2 UN No. : 2031

Synonyms : Aqua fortis, Azotic Acid General Use : Industrial chemicals

Manufacturer's Name : Deepak Fertilisers And Petrochemicals Corporation. Ltd.

Address: : Plot K-1, MIDC Indl Area, Taloja A.V., Dist: Raigad – 410 208

Telephone no.for info. : +91 - 022 - 67684000

SECTION 2 - COMPOSITION / INFORMATION ON INGREDIENTS

Composition : Nitric Acid 98% Hazardous : Nitric Acid

components

ACGIH TLV : 2 ppm

SECTION 3 - HAZARDS IDENTIFICATION

Primary Entry Routes: Inhalation, skin, eyes and ingestion

Acute Effect : Inhalation of vapours can cause breathing difficulties, severe

exposure may lead to pneumonia and pulmonary edema.

Ingestion can cause immediate pain & burns of mouth, throat and

gastrointestinal tract,

Skin contact can cause redness, pain and skin burns.

Carcinogenicity : Eye contact – vapours are irritating and may cause damage to

eyes.

Chronic Effect : Not listed as carcinogenic.

NFPA rating : Long term exposures seldom occur due to corrosive properties of

the acid, it may cause erosion of teach and lung damage.

Health -3, Flammability- 0

SECTION 4 - FIRST AID MESURES

Eyes : Immediately flush eyes with plenty of water for at least 15

minutes, lifting lower and upper lids occasionally get medical

attention immediately.

Skin: Remove contaminated clothing and shoes, flush skin with plenty



of water for at least 15 minutes, get medical attention

immediately.

Inhalation : Remove victim to fresh air. If not breathing give artificial

respiration, If breathing is difficult, give oxygen and get medical

Ingestion: attention immediately

Do not induce vomiting, give large quantities of water or milk if available; Never give anything by mouth if victim is unconscious,

or is convulsing. Obtain medical attention immediately.

Flash Point : Not Flammable.
Flash Point method : Not applicable

Auto-ignition : Not applicable.

Temperature : Not applicable.

LEL : Not applicable.

UEL : Not Flammable.Flammable classifn : It is not combus

Flammable classifn : It is not combustible, however, water spray may be used to keep fire exposed containers coal.

Extinguishing Media fire exposed containers cool.

SECTION 5 - FIRE FIGHTING MEASURES

Unusual fire or : It is not combustible but reacts with explosively with combustible organic or readily oxidisable materials, react with most metal to

release hydrogen gas.

Hazardous: Emits toxic nitrogen oxides fumes and hydrogen nitrate fumes

combustion Product and hydrogen nitrate when heated to decomposition.

Will react with water or steam to produce heat and toxic and

corrosive fumes.

Fire Fighting : Water spray may be used to keep fire exposed containers cool,

Instructions Ensure that water doesn't enter inside the containers.

SECTION 6 - ACCIDENTAL RELEASE MEASURES

Small spill : Shut off leaks without risk, dilute with alkali and drench with

Containment: water.

Clean Up : Prevent spillage from entering drains or water sources.

Dilute with alkali and wash with water.

SECTION 7 - HANDLING AND STORAGE

Handling Precautions: Protect from physical damage.

Storage Requirements : Store in a cool dry ventilated storage area with acid resistance

floors. Keep away from heat, water and incompatible materials.



SECTION 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

Engineering Controls Provide proper ventilation so as to maintain environment below

air borne exposure limit.

Respiratory If exposure limit is exceeded, use respiratory protection.

Protection

Protective Clothing / : Use full PVC Suit, PVC hand gloves and safety shoes.

equipment

SECTION 9 - PHYSICAL AND CHEMICAL PROPERTIES

Physical State Liquid

Colourless to pale yellow liquid, Chocking odour. Appearance & Odour :

Vapour Pressure 48 mmHg at 20°C

Specific Gravity 1.49 Water Solubility Soluble : (-) 42 $^{\circ}$ C Freezing Point : 84 °C **Boiling Point** Vapour Density 2 - 3

SECTION 10 - STABILITY AND REACTIVITY

Stability Stable under ordinary condition.

Chemical It is powerful oxidizing agent and is incompatible with strong incompatibilities

bases, metallic powder, carbides, hydrogen sulphide, turpentine and

combustible organics.

Light and heat. **Condition to Avoid**

Hazardous Emits toxic nitrogen oxides, fumes and hydrogen nitrate when

heated to decomposition. Decomposition

product

SECTION 11 - TOXICOLOGICAL INFORMATION

TLV as per ACIGH 2 ppm

Acute Inhalation Effect: Corrosive. Inhalation of vapour can cause breathing difficulties, over

exposure may lead to pneumonia and pulmonary edema.

SECTION 12 - ECOLOGICAL INFORMATION

Information not available

SECTION 13 - DISPOSAL CONSIDERATIONS

Disposal Instructions : It may be disposed off by neutralizing with alkaline materials and

water.

SECTION - 14 - TRANSPORT INFORMATION

Shipping Name : Nitric Acid

	DOT
Shipping name	Nitric Acid
Hazard class	8
UN Number	UN2031

SECTION 15 - REGULATORY INFORMATION

Oxidizing, Corrosive Material

SECTION 16 - OTHER INFORMATION

Prepared by : Deepak fertiliser and petrochemical corporation Ltd. Taloja

Disclaimer :

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